



Trouble Shooting

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I. Field Inspection

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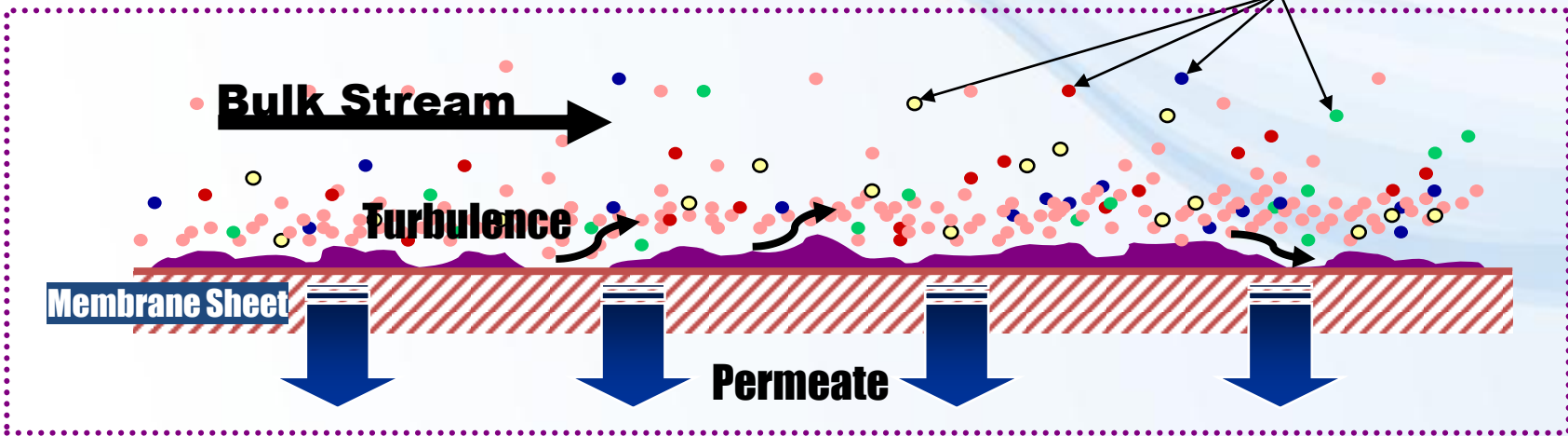
Loss of salt rejection, loss of permeate flow and pressure drop increase are the most common problems encountered in reverse osmosis.

Common Problems

- Loss in Flowrate
- Loss in Salt Rejection
- Increase in differential pressure

Causes

- Charged (+, -) and/or Non Charged Foulants
- Organic / Colloidal Matters
- Oxidizer / Debris
- Hydraulic Shock



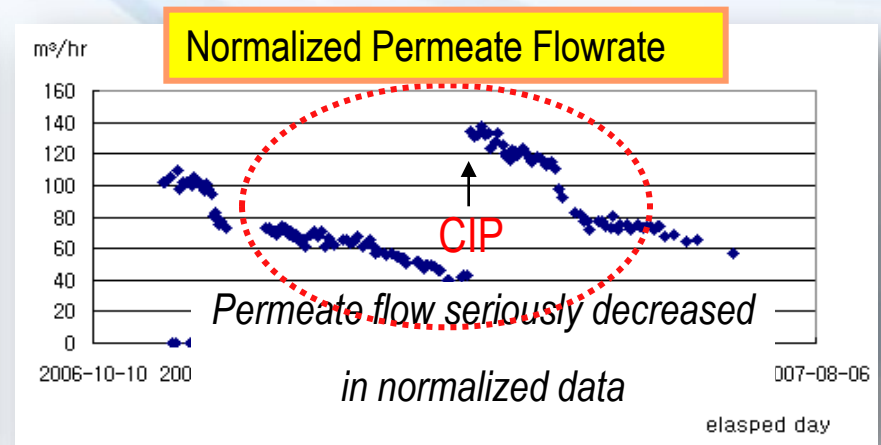
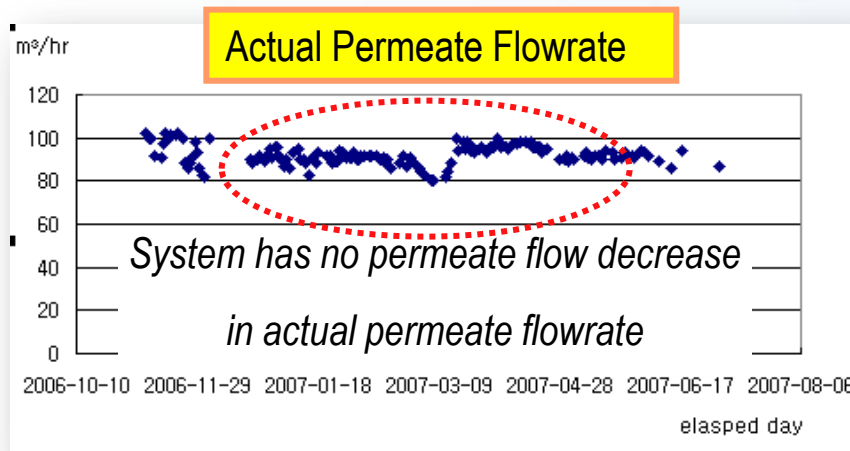
Evaluation of System

• Data Normalization

- Comparing real operating data to reference data (initial data) to predict membrane fouling and chemical damages, mechanical failure

• Normalization parameter

- Salt rejection ; Can be affected by pressure, temperature, recovery, etc.
- Flow rate ; Can be affected by pressure, temperature, pH, etc
- Differential pressure ; Can be affected by pressure and flow rate



This graphs are permeate flowrate trend before and after normalization

If the normalized performance show severe deterioration, the following should be checked.

- Calibration of all meters, sensors and pressure gauges
- Stabilization of system at least for 24 to 72 hours in continuous operation
- Consideration of permeate pressure
- Significant pressure loss from the feed to the concentrate
- Check of the start-up and shut-down procedure for hydraulic shocks, permeate backpressure and back-flow of permeate
- Check of CIP procedure and chemical
- Check of CIP frequency
- Water Analysis : Carbon dioxide can increase the permeate conductivity
- Check of the potential oxidation problem by chlorine and oxidizing chemicals
- Check of replacement rate of prefilters
- Check of the SDI
- Check of the scaling calculations and dosage rates of chemicals.



Check of Biogrowth and mold in tanks and pipes



Check of the vessels leakage which suck air during shut down and lead to hydraulic shock



Inspection of feed side by opening pressure vessel
: any fouling, smell, properly shimmed



Inspection of concentrate side by opening pressure vessel
: scaling



Check of couplers for torn, damaged or misplaced O-rings by removal the element from the pressure vessel



Inspection of element for fouling, scaling and mechanical damage

Localization of High Solute Passage

Profiling

If a system exhibits high solute passage, all individual vessel's TDS or conductivity or other relevant quality values must be checked.

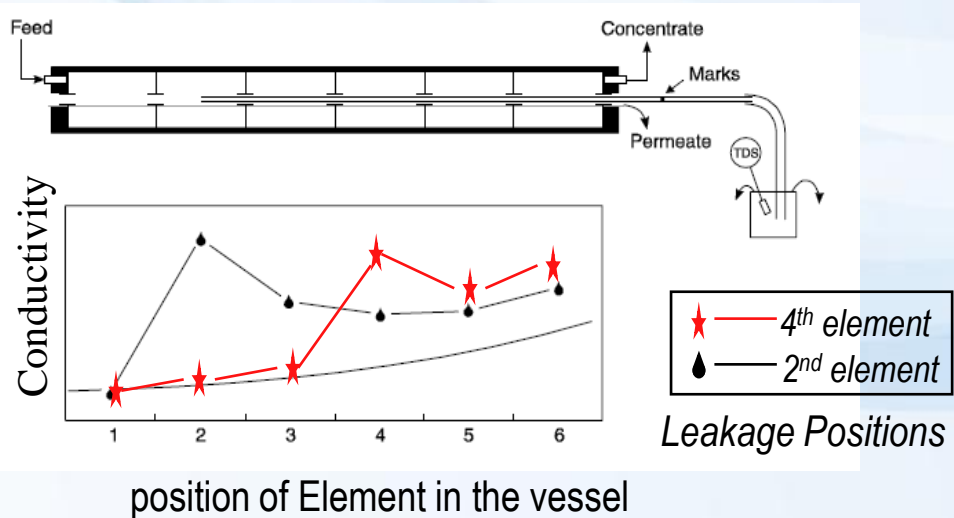
Probing

If one pressure vessel shows a significantly higher permeate concentration than the other vessels of the same stage, conductivity of every elements need to be checked

Profiling



Probing



A close-up photograph of water splashing, showing droplets and ripples, positioned at the top of the slide.

II. Membrane Analysis

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Visual inspection provides *some information on problems like fouling and scale.*



Check Point

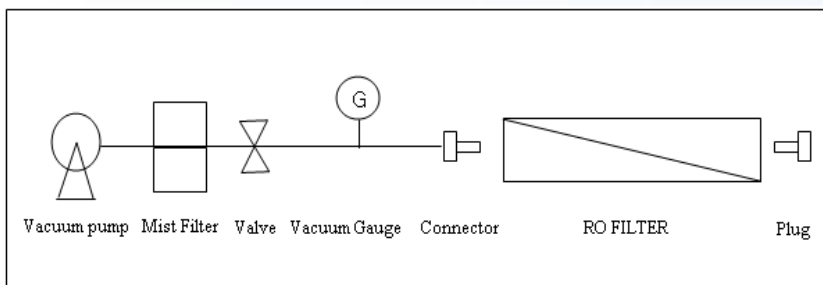
- Potential fouling or scaling problem
- Discolorations of the outer wrapping and the fiberglass
- Any deposits or foreign matter
- Telescoping, channeling and fiberglass damage
- Mechanical damage of permeate water tube

Vacuum Test is performed *to detect any mechanical leaks* by checking the vacuum decay rate. (Based on San Diego Protocol or ASTM)



Check Point

- Leaks by puncture or scratches
 - Delamination and physical damage by permeate back pressure or water hammer
 - Membrane integrity
- ❖ Test acceptance criteria
- vacuum decay rate < 100 mbar/min.



Vacuum test equipment

II. Membrane Analysis Performance Test

Standard element performance test is used to confirm the *performance comparison with initial value* under standard test conditions



Table 1. Performance Data on Returned Membrane

Bar code	Catalogue Value		Re-tested value			VDR (mbar/min)	Weight (kg)	Remarks
	Flowrate (GPD)	Rejection (%)	Flowrate (GPD)	Rejection (%)	DP (kgf/cm ²)			
BJFAA0110032	11,000	99.5	16,700	98.3	0.5	20	15.8	1st stage A42-1

Initial differential pressure is 0.3kgf/cm²

※ Test conditions
 Feed Concentration : 2,000 ppm NaCl
 Feed Pressure : 225 psi
 Temperature : 25°C
 Recovery : 15%
 pH : 6.5

Check Point

- Comparison of salt rejection, permeate flowrate and differential pressure increase with initial data
- Performance comparison before and after any cleaning test

II. Membrane Analysis Flat Sheet Test

Flat sheet test is performed to **check the performance by parts**

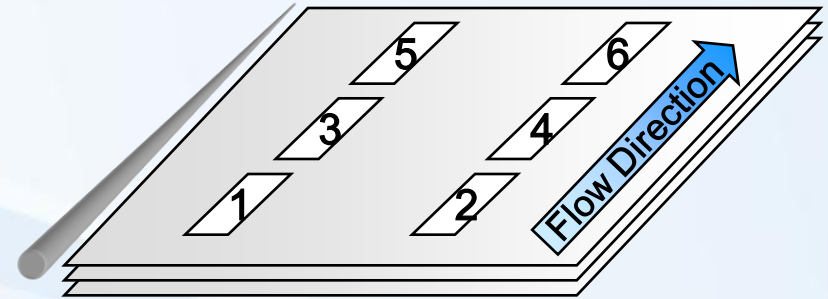


Figure. Membrane Sampling Points & Examples

Check Point

- Comparison of salt rejection and permeate by location
- Performance comparison before and after any cleaning test

Table 2. Flat Sheet Performance Test

Sample	Flux (GFD)	Rejection (%)	Remarks
1	50.2	93.21	Feed part
2	46.2	94.92	
3	test error		Middle Part
4	48.5	95.68	
5	47.2	97.33	Concentrate Part
6	46.6	97.83	
average	47.7	95.79	

To confirm **whether bio-fouling or scaling is the cause of decline** in membrane performance, either acid or caustic solution soaking methods are used.

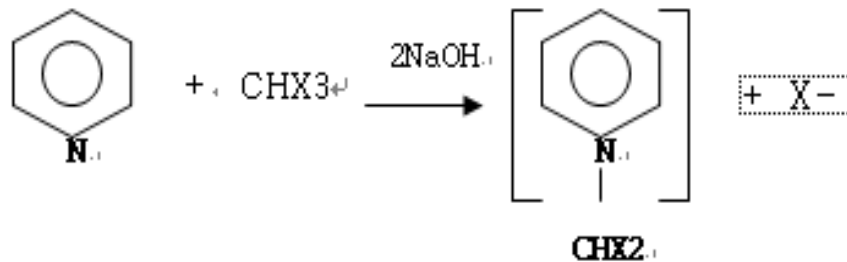
The image displays six pairs of petri dishes, each representing a different cleaning agent used to remove fouling from a membrane. Each pair shows the membrane before and after treatment. The cleaning agents and their concentrations are as follows:

- 0.1% STP
- 0.2% HCl
- 1% EDTA
- 2% Citric acid
- 0.03% SDS
- 0.1% NaOH (highlighted with a red border)

Fujiwara test method is to confirm qualitatively whether the PA polymer structure was attacked by oxidative halogen chemicals like chlorine, bromine and iodine or not

Only oxidized RO membrane turns pink by this test.

- ① 1cm² membrane is placed in the bottom of tube
- ② Several milliliters of 5N NaOH are added
- ③ Followed by 0.5mL of Pyridine
- ④ Test tube is placed in a beaker of boiling water



Comparison with normal membrane

To determine the causes and the location of a salt passage, the element is operated with a pressurized dye solution prior to an autopsy. A pink permeate would indicate membrane damage.

Check Point

- Oxidation and damage of membrane
- Permeate Back-Pressure



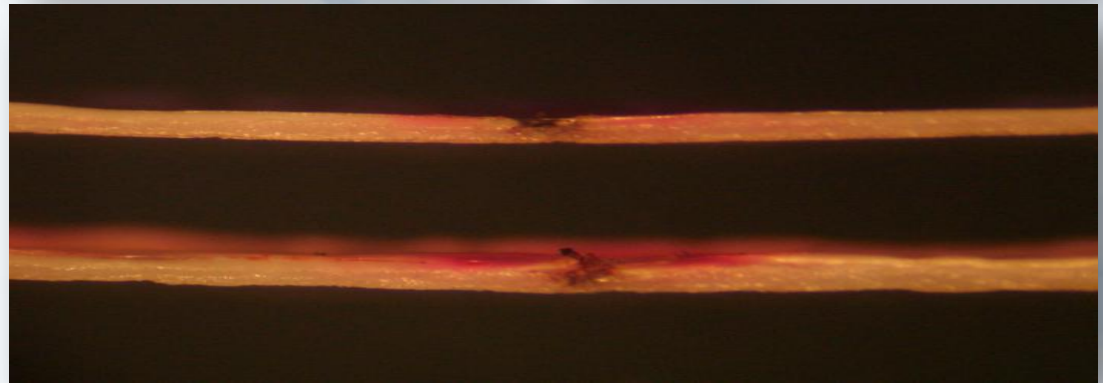
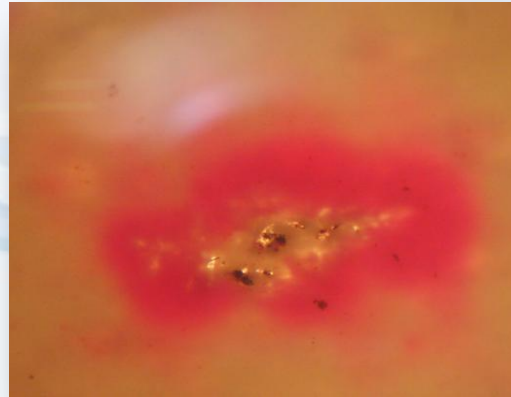
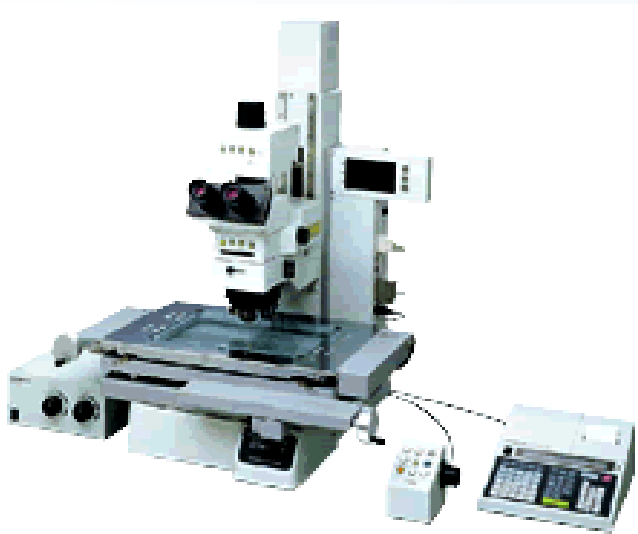
*Rhodamine B is used for dye test.
-> Rhodamine B's M.W. is 400 ~500 and can not pass Non-Damaged Membrane.*



Dye solution passed through the damaged part of concentrate side.

II. Membrane Analysis **Optical Microscope**

To verify the causes and location of decline in performance, microscope inspection is performed.



Check Point

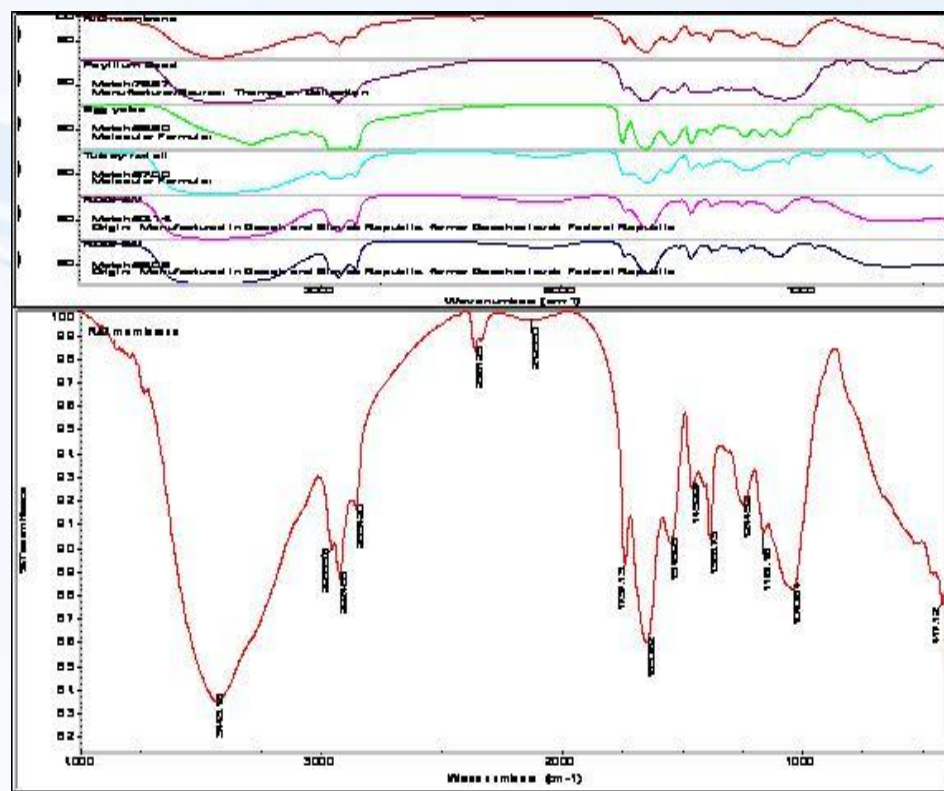
- Fouling Particles
- Membrane Damage Part

Above pictures show damaged membrane by some welding particles

Specific organic / inorganic matters are confirmed by FT-IR (Fourier Transform Infrared Spectroscopy)

Check Point

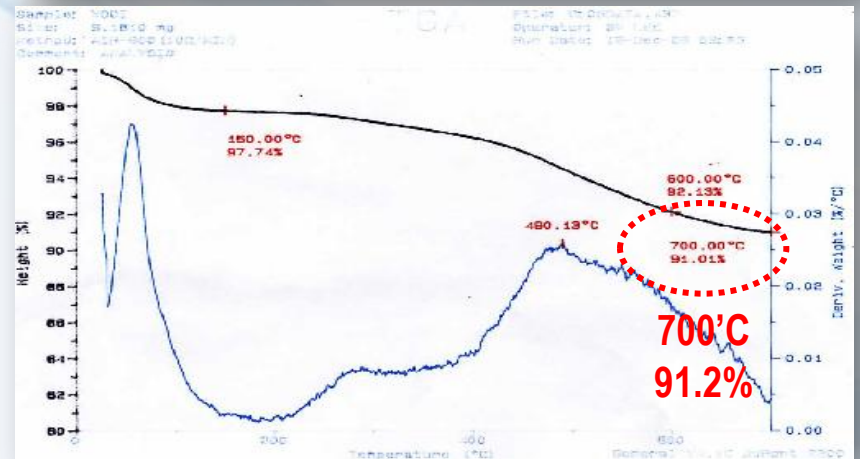
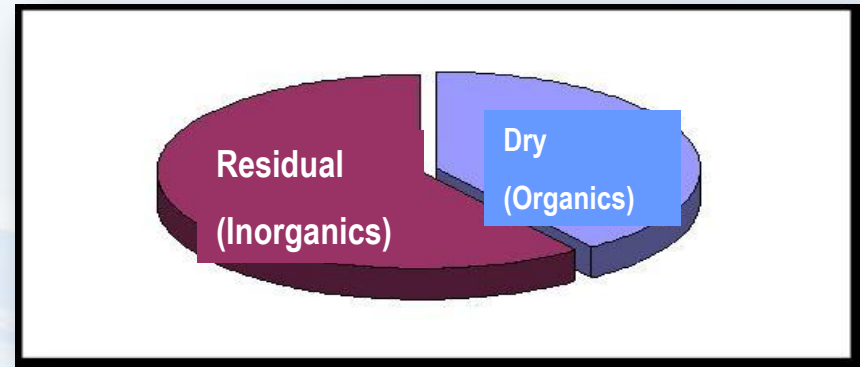
- Infrared radiation wave indicate property of organic matter functional group
- Through the comparison of spectrum of matter, possible to confirm pollutants



The content rate of organic/inorganic matter is verified.

Residual rate indicates inorganic ratio of foulants.

- ① Weight calibration of crucible
- ② Weight calibration of crucible and sample
- ③ Heating of crucible and sample
- ④ Weight calibration of crucible and ash
- ⑤ Calculation of organic matter : [②- ④]



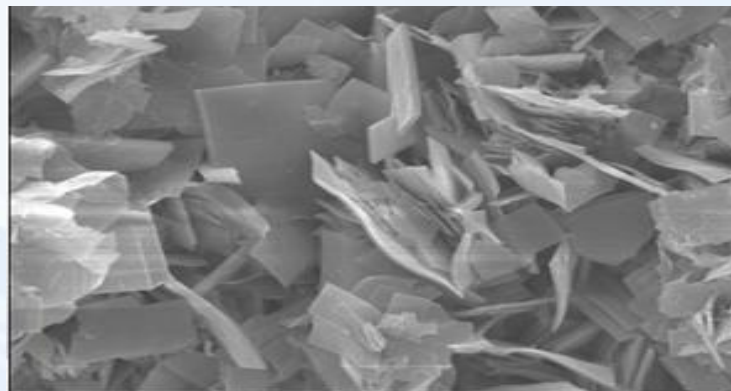
This figure shows 90% of most foulants are resided and these are inorganic matters.

SEM (Scanning Electron Microscope)

The sample is **examined under a microscope at various magnifications to verify some foulants.**

Check Point

- The structure
- Crystallization
- Size and kind of the membrane foulants



SEM image of CaCO3 Scale



SEM image of micro-organism

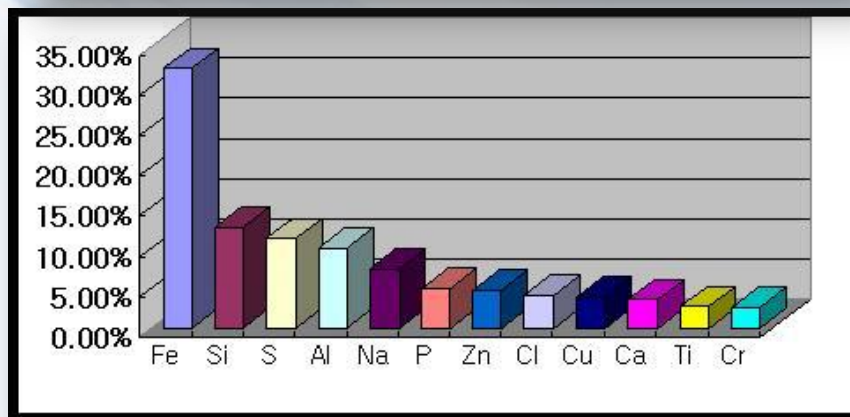
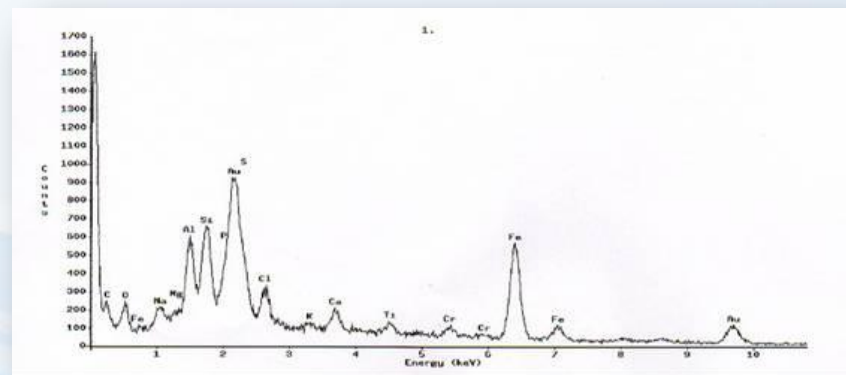


EDS (Energy Dispersive Spectrometer)

EDS gives a *semi-quantitative composition of chemical elements of the membrane and the foulants.*

Check Point

- Proportion between energy intensity and content of matters (Semi-quantitative)
- Useful to choose CIP method



Fe is main fouling component, this value indicate semi-quantitative of fouling matters.

A dynamic splash of clear water against a light blue background, with droplets and ripples visible.

III. Trouble Shooting Guide

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Metal Oxide Fouling

III. Trouble Shooting guide



Symptoms

- ✓ Normalized **DP Increases** Especially in 1st Stage
- ✓ Normalized Permeate **Flow Decreases**
- ✓ Normalized **Salt Rejection Decreases**

Corrective Methods

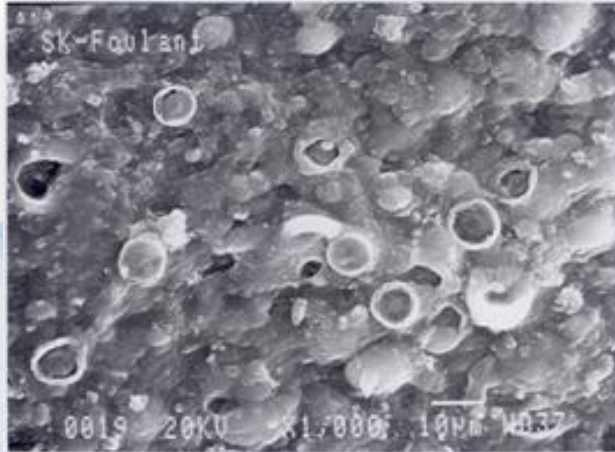
- ✓ Clean the membrane elements
- ✓ Adjust, correct and /or modify the pretreatment
- ✓ Retrofit piping or system components with appropriate materials.

Verification Method

- ✓ Analyze feedwater for iron and aluminium
- ✓ Check system components for evidence of corrosion

Biological Fouling

III. Trouble Shooting guide



Symptoms

- ✓ Normalized DP Increases at Any Stage
- ✓ Normalized Permeate Flow, Salt Rejection decrease

Corrective Methods

- ✓ Clean and sanitize the entire system
- ✓ Soak and rinse in High pH
- ✓ Installation of fouling resistance elements

Verification Method

- ✓ Bacteria Count : TBC, DBC
- ✓ SEM Analysis



Colloidal Fouling

III. Trouble Shooting guide



Symptoms

- ✓ Normalized DP Increases Especially in 1st Stage.
- ✓ Normalized Permeate Flow, Salt Rejection decreases

Corrective Methods

- ✓ Clean the elements depending on foulants
- ✓ Adjust, correct and /or modify the pretreatment

Verification Method

- ✓ Check the feed water SDI
- ✓ Analyze accumulations on prefilter cartridges
- ✓ Inspect and analyze deposits on elements



Symptoms

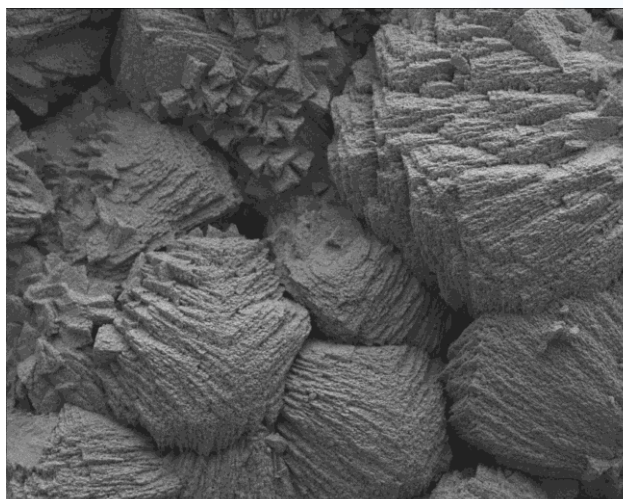
- ✓ Normalized DP Increases Especially in last Stage.
- ✓ Normalized Permeate Flow, Salt Rejection decreases

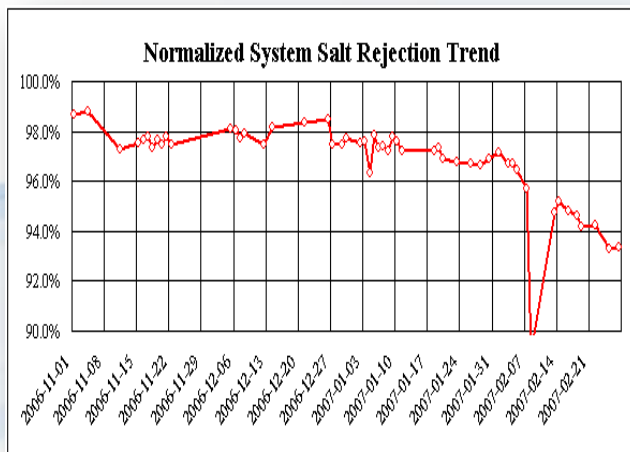
Corrective Methods

- ✓ Pretreatment : Softening, Lime softening, Acid injection
- ✓ Optimize cleaning depending on present scaling salts
- ✓ Anti-scalant Injection : SHMP, Organo phosphate etc

Verification Method

- ✓ Check feedwater analysis for the scaling potential at system recovery
- ✓ Inspect concentrate side of system : weigh and touch the tail-element





Symptoms

- ✓ Normalized Salt Rejection Decrease
- ✓ Normalized Permeate Flow Increase
- ✓ Normalized DP is Stable

Corrective Methods

- ✓ Monitoring Free chlorine, ORP value
- ✓ Injecting SBS
- ✓ Chloramine disinfection instead of chlorine

Verification Method

- ✓ Dye Test
- ✓ Fujiwara Test
- ✓ ESCA Analysis



Conductivity increase of several vessel by O-ring leak.

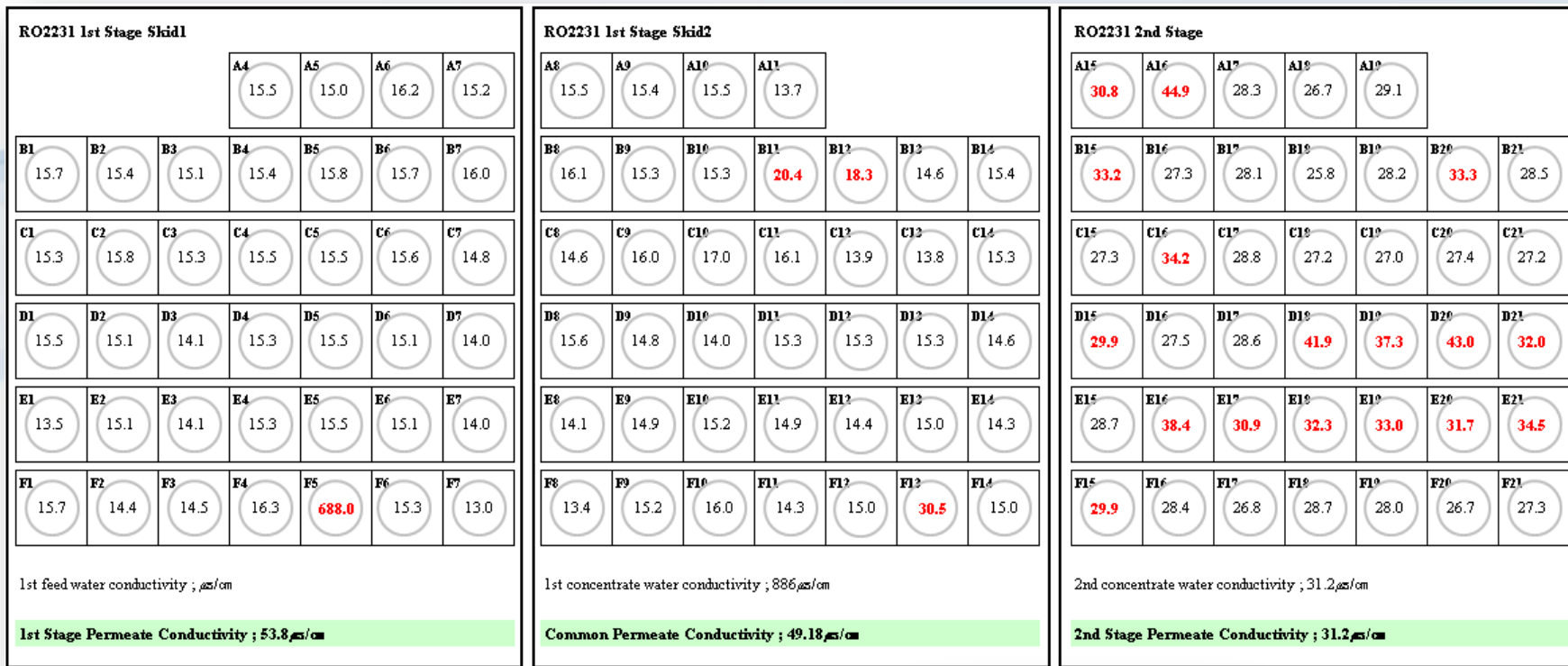
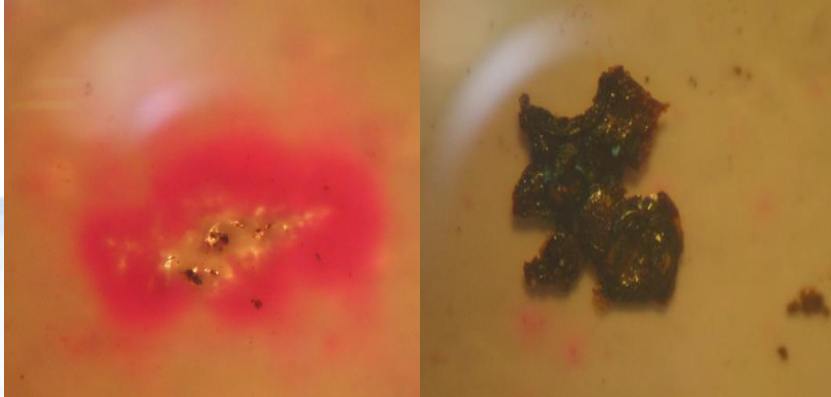


Figure. Profiling for each vessel



Cause broken o-ring

Mechanical Damage by Welding Debris



Symptoms

- ✓ Normalized Salt Rejection Decrease

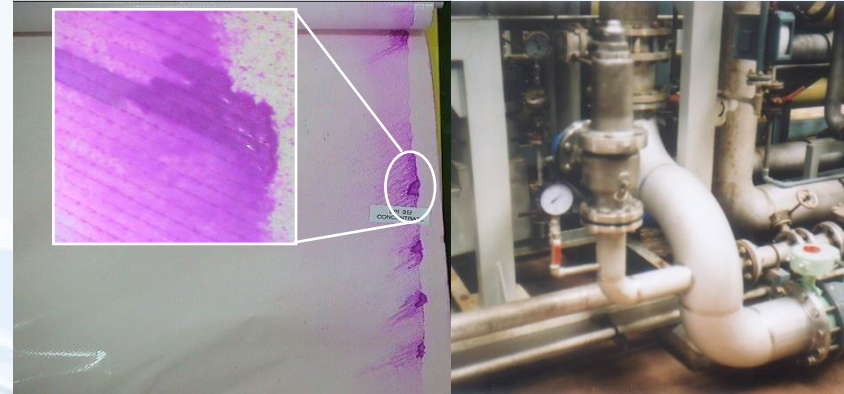
Corrective Methods

- ✓ Enough system flushing before installation

Verification Method

- ✓ Optical Microscope

Permeate Back Pressure



Symptoms

- ✓ Normalized Salt Rejection Decrease

Corrective Methods

- ✓ Check valve in the permeate pipe inspection
- ✓ Installation Pressure Release Valve In Permeate Line

Verification Method

- ✓ Dye Test / Probing

Symptoms		Causes
Low Permeate Flow	First Stage	Deposition of organic/colloidal matter Initial biofouling / Metal Oxide Fouling
	Last stage	Scaling
	Others	Aged Preservation Solution While Stocked Compaction and Intrusion Incomplete Wetting
High Salt passage	overall	Membrane Oxidation Membrane Surface Abrasion
	specified	O-rings leakage / Telescoping Permeate Backpressure
High differential Pressure		By pass in Cartridge Filter Pretreatment Media Filter Breakthrough Pump Impeller Deterioration /Precipitated Antiscalants Brine Seal Issues Biological Fouling / Scaling

A dynamic splash of clear water against a light blue background, with droplets and ripples visible.

IV. CIP (Clean In Place)

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CIP is, spelled 'Clean In Place', cleaning membranes in the RO vessels with equipped, not change an arrangement.

Elements should be cleaned immediately when one of the following symptoms is detected.

- ✓ Normalized water flow has decreased by 10-15% from start-up conditions.
- ✓ Delta P, or pressure drop over a stage or the system, has increased by 10-15%.
- ✓ Salt rejection has decreased (ie permeate TDS has increased) significantly over time.

Even if the above symptoms don't show, it is good to do CIP every 1-2 times per year for preventing some problems.

Inorganic scale

IV.

CIP(Clean In Place)

The main cause of inorganic scale is the concentration increase due to high recovery on RO operation.

Scale	Scale with Ca & Mg	CaSO ₄	SiO ₂	Fe & Al	Remark
Cause	The concentration increase due to high recovery			Corrosion of pipe	
Location	Tail vessel & elements			Overall	
Cleaning chemical	Strong Acid	Chelate	None	Weak Acid	
pH	2.0~	10.5~12.0	-	3.5	
Temp.	30 ~35 °C	30 ~35 °C	-	25 °C	

- ✓ In case of SiO₂, there are not effective chemicals particularly. Decreasing recovery & pre-treatment for SiO₂ are important.
- ✓ In case of Fe, it is necessary to remove Fe in pretreatment because oxidizing would be stronger if the oxidizer meet Fe.

Organic fouling

IV.

CIP(Clean In Place)

The main cause of organic fouling is the concentration increase of fouling on membrane surface on RO operation.

Fouling	Organic compound	Bio-Film	Silt & colloid	Remark
Cause	The concentration increase of fouling on membrane surface			
Location	Lead vessel & elements			
Cleaning chemical	Surfactant	Alkali+Surfactant	Chelate	
pH	10.5~12.0	10.5~12.0	10.5~12.0	
Temp.	30 ~45 °C	30 ~45 °C	30 ~45 °C	

- ✓ Bio-film occurs due to microbial synecology. This can be suppressed by injecting a biocide prior to RO system. However, oxidizing biocide for PA should be prohibited or neutralized before RO feed stream.

CIP Procedure

CIP procedure is determined by feedwater component, trouble symptom and field condition. The engineer's experience is the most important factor.

The following is a common procedure of CIP

Prepare the cleaning solution per the instructions

- Temp. : up to 35°C (upper 35 °C, change in PA properties is concerned)
- pH : Acid 1.5~2.5, Alkali 11~12

Circulation overall system - 10~60 min.

Circulation 1st, 2nd, 3rd bank : each 10~60 min.

Soaking : 4~15 hours (max. 3 days)

Re-circulation 1st, 2nd, 3rd bank : each 10~60 min.

Low pressure flushing : each 10~60 min.

High pressure flushing : until obtain a target permeate quality

- ✓ Circulation - Feed pressure : 1.4~4.1 bar, Feed flow each vessel(8") : 30-40gpm
- ✓ The size of CIP Tank = all pipes volume(feed+return) + all vessels volume

Thank You!

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